

ReuniãoAnual do INCT-FCx - 2022

De 26 a 29/Outubro de 2022, Hotel Estância Atibainha, Nazaré Paulista/SP

## AN IMIDAZOLE-CONTAINING NOVEL CATIONIC SURFACTANT

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Abstract: Vesicles formed by cationic surfactants are used for gene and drug delivery and recently for vaccines. Imidazole-derived surface-active ionic liquid derivatives have many applications in nanotechnology and biotechnology (1). Here we synthesized and characterized a novel cationic surfactant containing an imidazolium head group and two alkyl chains and compared it with a similar compound. 1,3-Di-hexadecyl-imidazolium chloride (DHImC) was synthesized and characterized. Dihexadecyldimethyl-ammonium chloride (DHDAC) was prepared by ion exchange from commercial DHDAB. 4iodide trimethylammonium-2,2,6,6-tetramethylpiperidine-1-oxyl (CAT) was encapsulated in SUVs of both surfactants in 5 mM NaCl (extruded with 100 nm pore membrane). EPR spectra for 5-MeSL and 16-MeSL (Methyl n-DOXYL-stearate) were incorporated in the surfactant bilayer. Differential Scanning Calorimetry determined transition temperatures (Tc). Optical Microscopy images were obtained for Giant Unilamellar Vesicles (GUVs). The Tc's were: 30.1 °C (DHDAC) and 48.7 °C (DHImC). CAT-EPR spectra of SUVs, recorded after adding Ascorbic Acid (AA), showed a stable residual EPR signal attributed to aqueous core-entrapped CAT. 5- and 16- MeSL spin labels showed a more ordered region closer to the polar head (5-MeSL) and a less ordered region at the hydrophobic core of the bilayer (16-MeSL). The mobilities at different temperatures reflected a gel phase before the Tc. Optical Microscopy images also showed a clear difference in vesicle organization before and after the transition temperature. Conclusions. DHImC forms stable vesicles with an aqueous inner compartment. DHDAC and DHImC, characterized by EPR, showed differences in mobility along the bilayer and Tc's. The remarkable differences in Tc's for surfactants with the same alkyl chain size can be explained by the the  $\pi$ - $\pi$  head group interactions on the imidazolium compounds. DHImC vesicles constitute a novel and potentially useful delivery system.

**Key-words**: Imidazol-surfactant, Micelles, bilayer **Support:** This work has been supported by FAPESP, CNPq, INCT-FCx, NAP-FCx. **References:** 

[1] Kanjilal, S. et al European J. Lipid Sci. Technol.111, 941 (2009).